



Louisiana Alliance for Simulation-Guided Materials Applications

# Temperature Dependence of Porosity for Porous Electrolyte Based $\text{No}_x$ Sensors

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# Outline

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- Acknowledgements

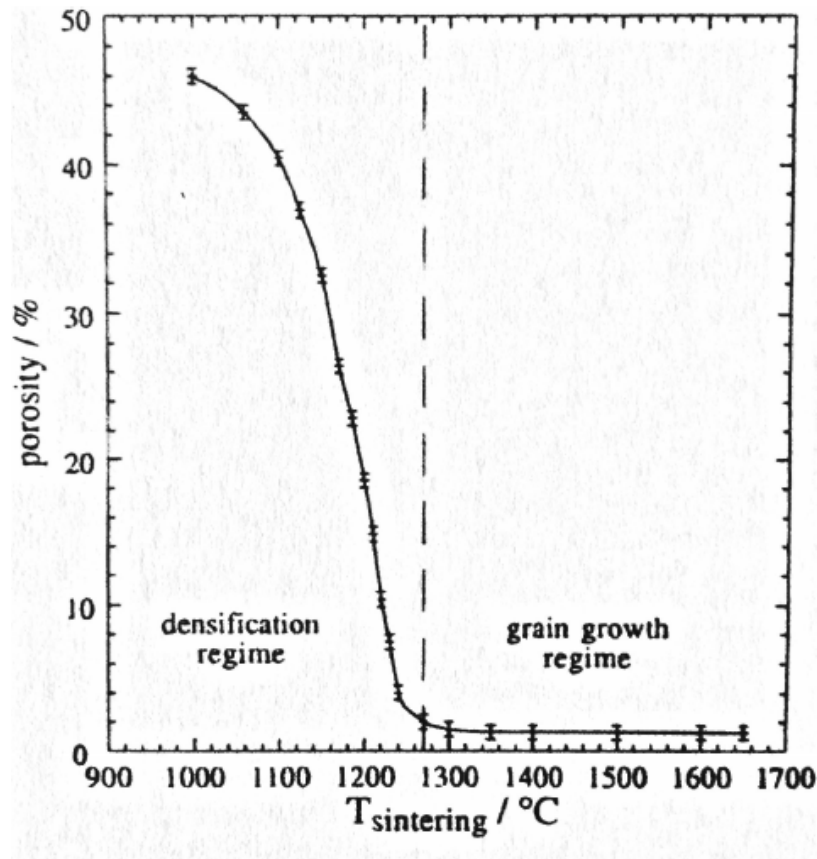
# Introduction

- Goal: To develop more sensitive emission sensors for automotive exhaust systems
- NO and NO<sub>2</sub>, collectively described as NO<sub>x</sub>, are the primary pollutants
  - Emissions and fuel economy controlled by the same system

# NO<sub>x</sub> Sensors

- Conventional sensors
  - Made of Yttria-Stabalized Zicronia (YSZ)
  - Operate at high temperatures
  - Dense microstructure
- Experimental sensors
  - Made of YSZ
  - Operate at high temperatures
  - Porous microstructure
  - Sensitivity depends on porosity
  - Porosity depends on fabrication temperature

# Project Motivation

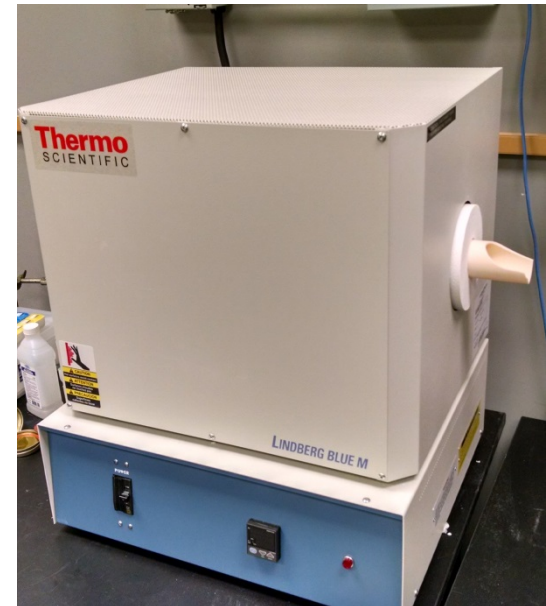


- We want to determine the porosity that yields the optimum sensitivity
  - i. Cannot be too porous
  - ii. Cannot be too dense
- Begin by investigating how pure 8% molar Tosoh™ YSZ behaves with temperature
  - i. Steil and Thevenot<sup>1</sup> showed that porosity will decrease with temperature as seen in figure 1.

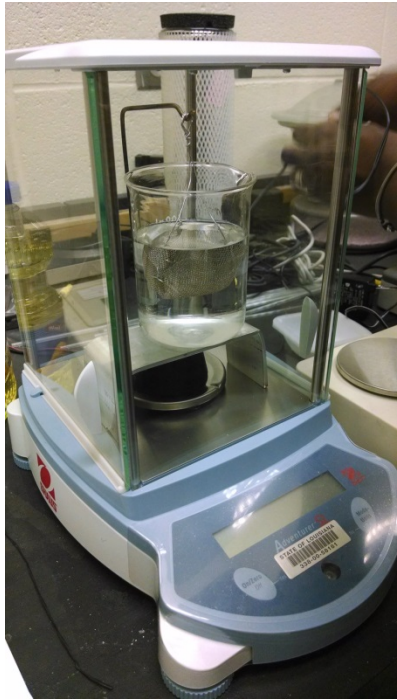
Figure 1. Porosity vs. Sintering Temperature

# Sample Making

- 500 mg of YSZ was placed in a die press and pressed at approximately 200 Mpa
- Samples were fired at 950°C, 1000°C, 1050°C, 1100°C, 1150°C, and 1200°C for one hour.
- 5 Samples were made per temperature
  - 3 used for Archimedes porosity measurements
  - 1 for Scanning Electron Microscopy (SEM) imaging
  - 1 spare
- Mass at fabrication and press pressure have negligible effects on final porosity



# Archimedes Porosity Measurements



- The Archimedes Principle was used to determine porosity using the following formula

$$Porosity = \frac{W_{\downarrow wet} - W_{\downarrow dry}}{W_{\downarrow displaced}}$$

- Samples were soaked for approximately 20 hours to saturate pores
  - Samples were soaked for 1 hour between runs
- 3 Archimedes measurements were then taken at one hour intervals

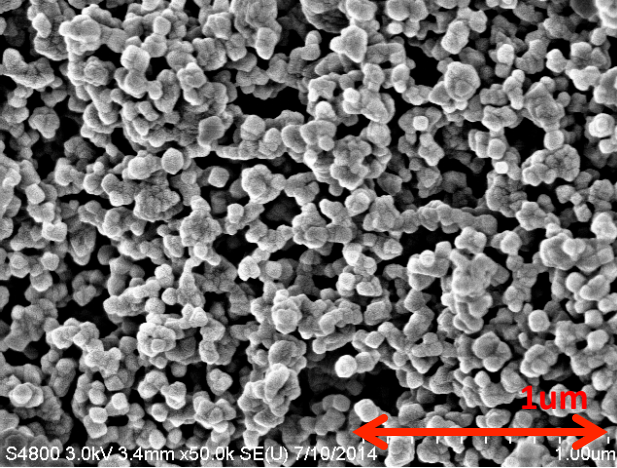


# SEM Imaging

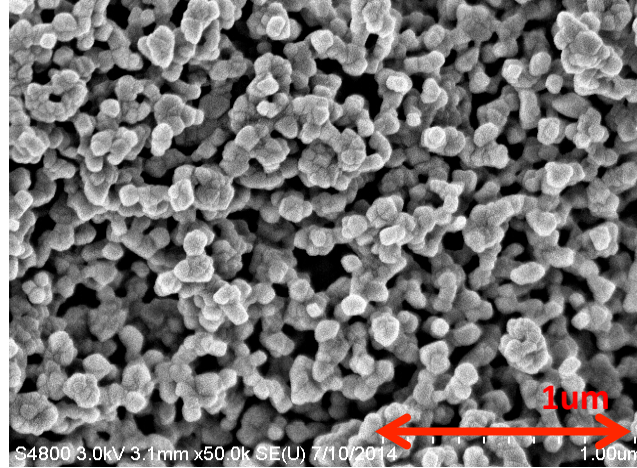
- Samples had to be gold plated
- SEM images were taken at 3 different points for each sample
  - Each point had three different magnifications
  - Images intended to be representative (devoid of cracks, abnormal growths, etc. . .)
- A total of 54 pictures were taken and used for analysis



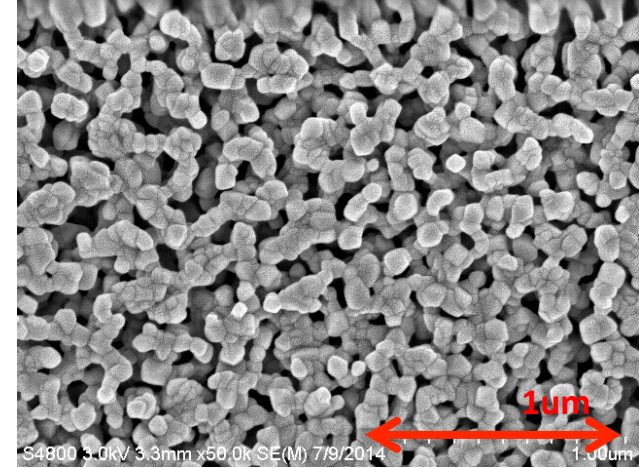




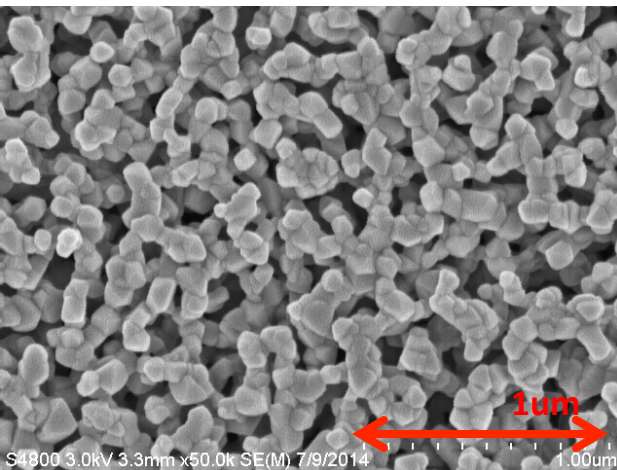
a



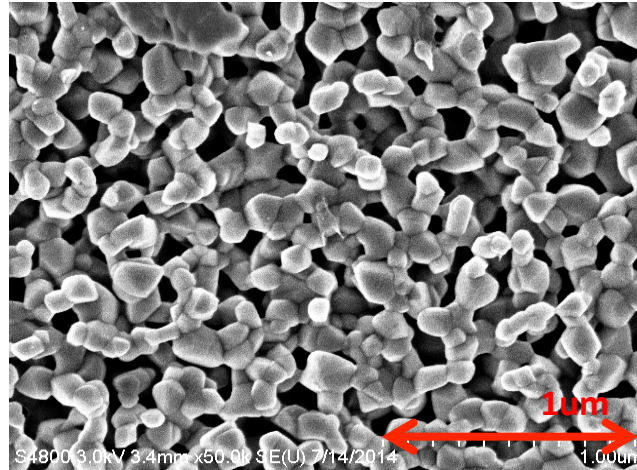
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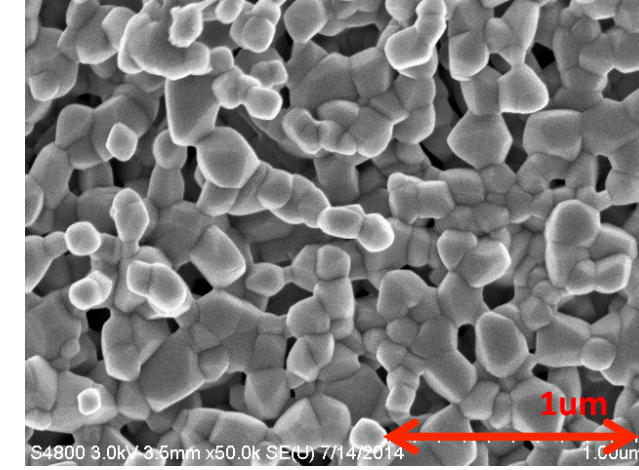
c



d



e



f

Figure \_: Particle Coalescence for a.) 950°C b.)1000°C c.)1050°C  
d.)1100°C e.)1150°C f.)1200°C

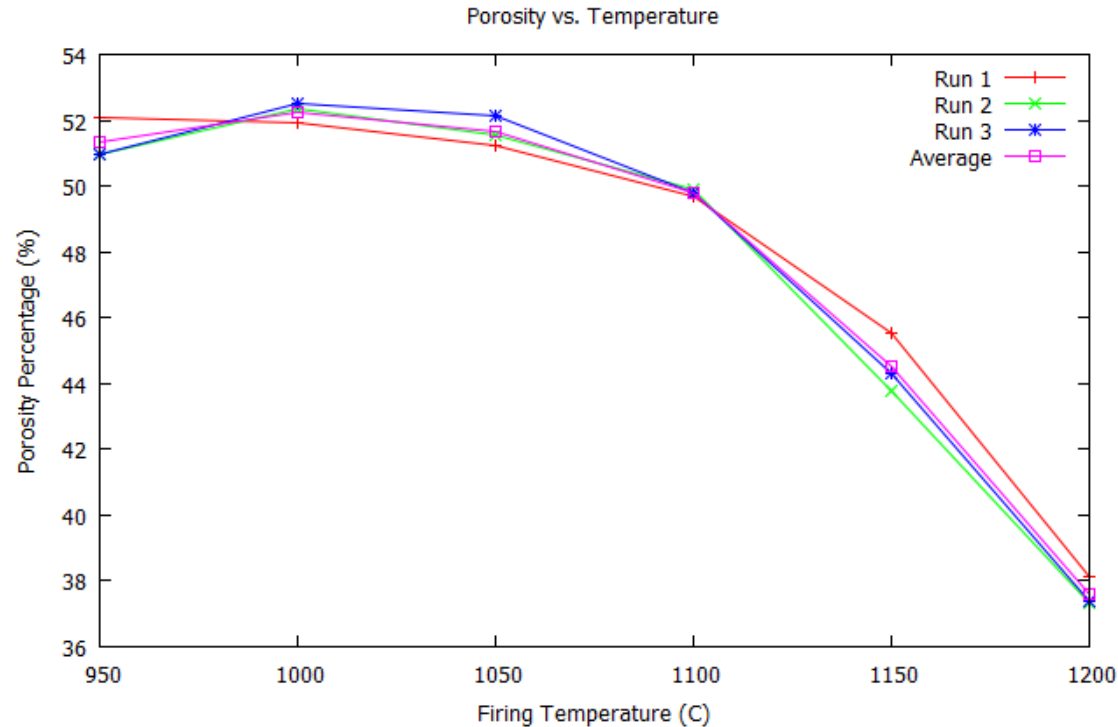
# Image Processing and Analysis

- Use MATLAB to analyze SEM images
  - Each pixel is assigned a brightness value between 0-255, 255 being white
  - Numerically integrate brightness values using the following formula

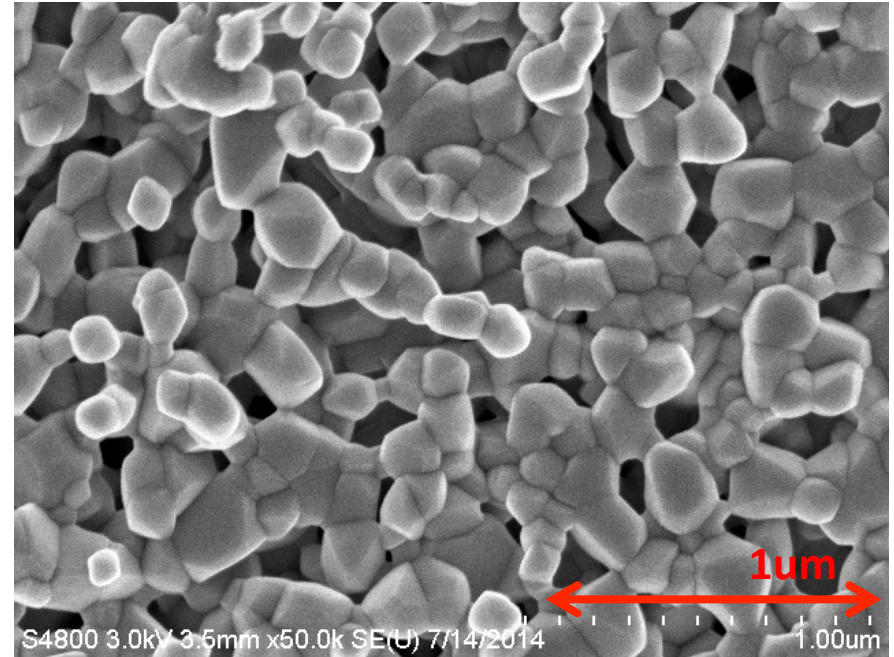
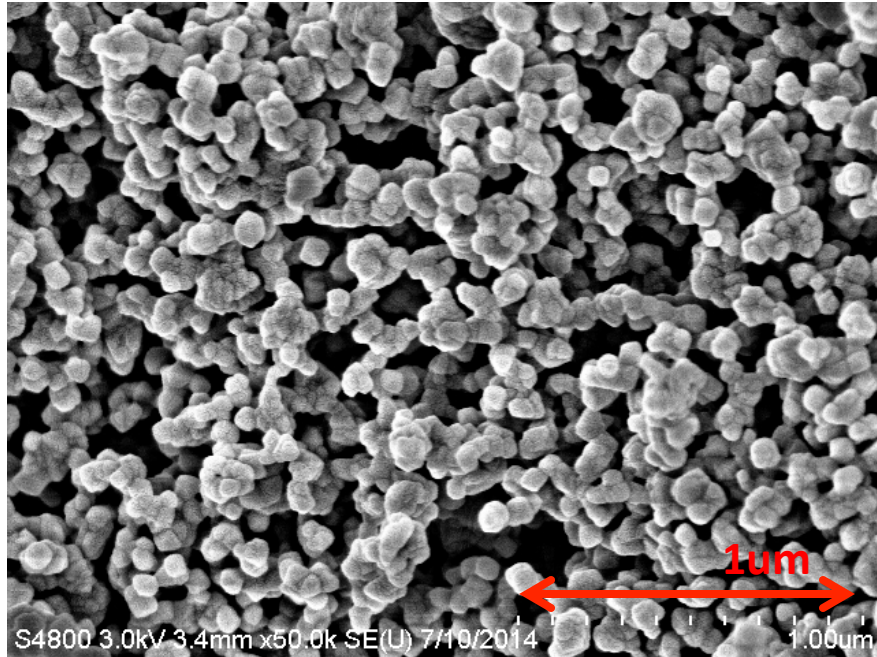
$$Porosity \approx 1 - \frac{\sum_{i=1}^{nx} \sum_{j=1}^{ny} f(x_i, y_j) - f_{\min} (nx\Delta x)(ny\Delta y)}{(f_{\max} - f_{\min})(nx)(ny)}$$

# Archimedes Results

- Archimedes results agree with the trend predicted by Stein and Thevenot.
  - 3 runs of Archimedes observed
  - Differences between this experiment and theory can be accounted for with how the samples were made
  - Runs 2 and 3 demonstrate error at 950°C. This is most likely due to how brittle the sample is



# Image Processing Results



Temperature (°C)	SEM Porosity %	Archimedes Porosity %
950	50.65	52.09
1200	44.36	38.13

# Summary

- Samples were made with 500mg of YSZ
  - Fired at 950°C, 1000°C, 1050°C, 1100°C, 1150°C, and 1200°C for one hour.
- Archimedes results agree with results shown by Steil and Thevenot
- Simulation results appear to agree with Archimedes experiments.

# Acknowledgements

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- **Thank you to Dr. Naidu Seetala for his guidance in the Archimedes Porosity Measurement experimental set up**
  - **Thank you to Louisiana Tech University and to LA-SIGMA for allowing me this opportunity.**

# References

<sup>1</sup>Ling, C., Han F., Dai, W. and Murray, E.P. "Influence of Microstructure on the Sensing Behavior of NOx Exhaust Gas Sensors" *Journal of the Electrochemical Society*. Volume 161, Issue 3 (2014): B34-B38

<sup>2</sup>Steil, M. C. and Thevenot, F. "Densification of Ytria-Stabilized Zirconia." *Journal of the Electrochemical Society*. Volume 144, Issue 1 (1997): 390-398