### Semiconductors

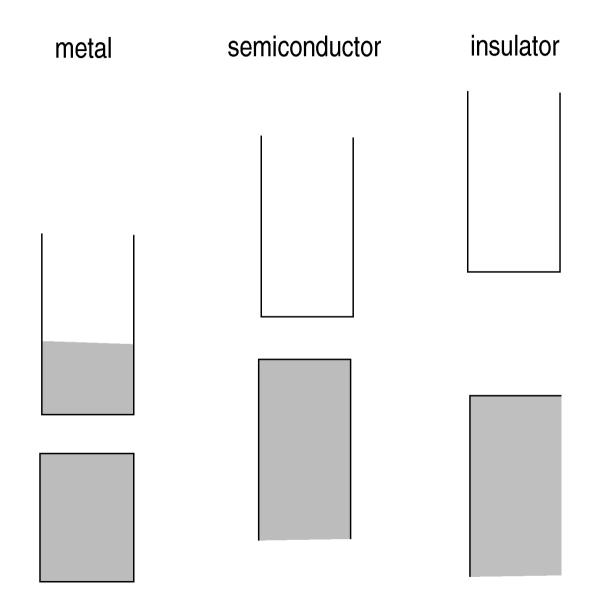


# **Walter Schottky** (1886-1976)

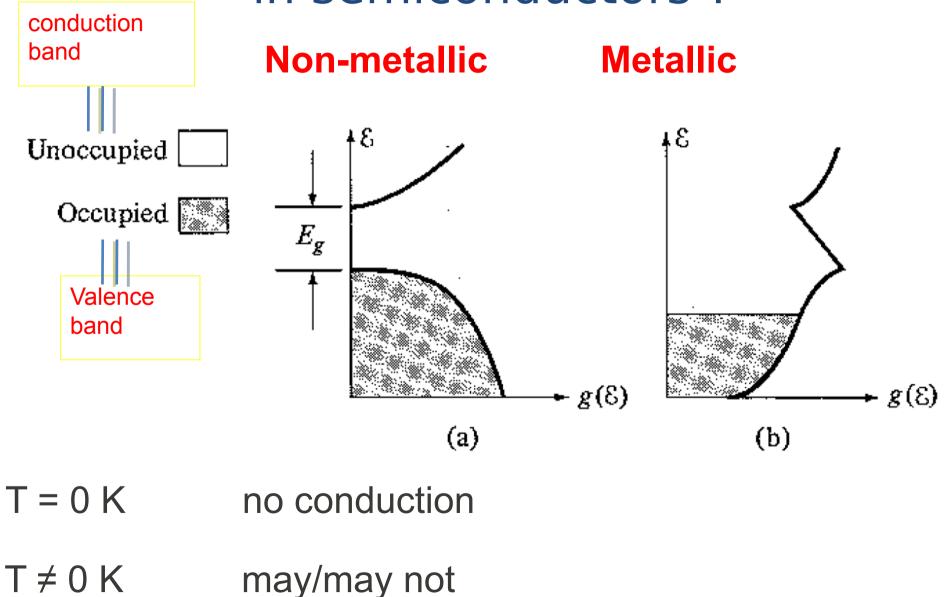


# John Bardeen (1908-1991)

### **Band Structure**



# Where are the carriers coming from in semiconductors ?



# What is a Semiconductor?

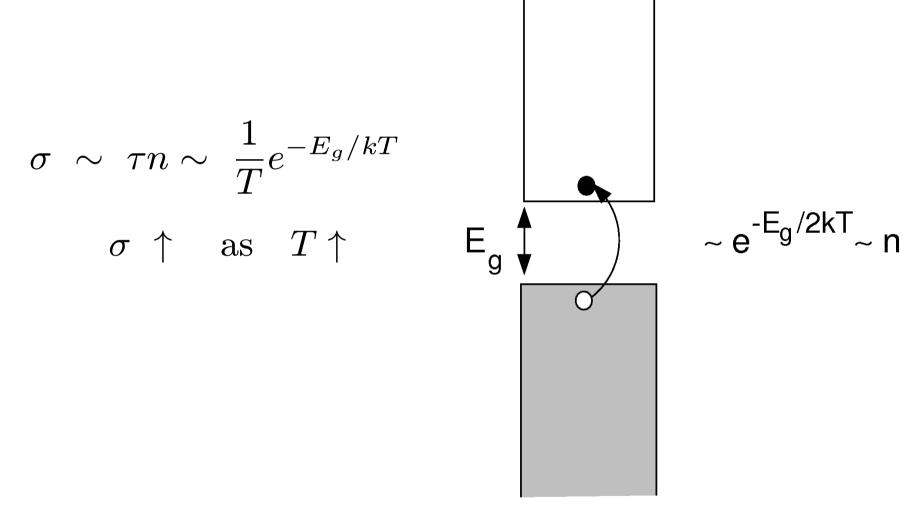
Most condensed matter physicists make the distinction on the basis of the conductivity and its temperature dependence. In the Drude model (parabolic band)

$$\sigma = \frac{ne^2\tau}{m^*}, \qquad \mu = \frac{e\tau}{m^*}, \qquad \sigma = ne\tau$$

$$\longrightarrow \text{ metals} \qquad \sigma \sim \tau \sim \frac{1}{T}, \qquad \sigma \downarrow \text{ as } T \uparrow$$

0

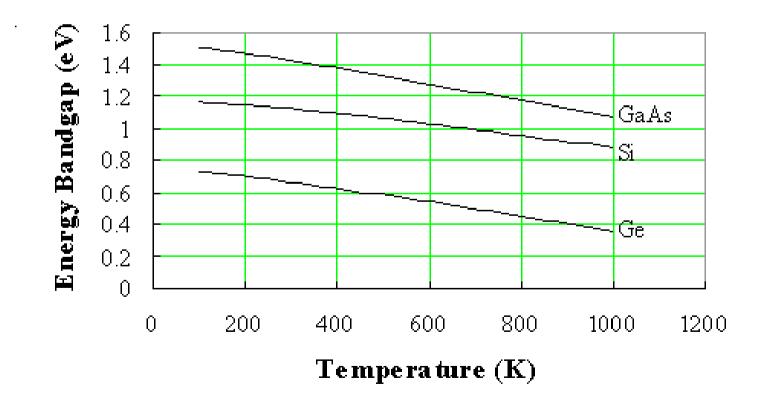
In semiconductors, the population of free carriers n is temperature dependent. The exponential always will dominate the power law dependence of  $\tau$ .



The same is true for insulators, of course, except here n is so small that for all realistic purposes  $\sigma \sim 0$ .

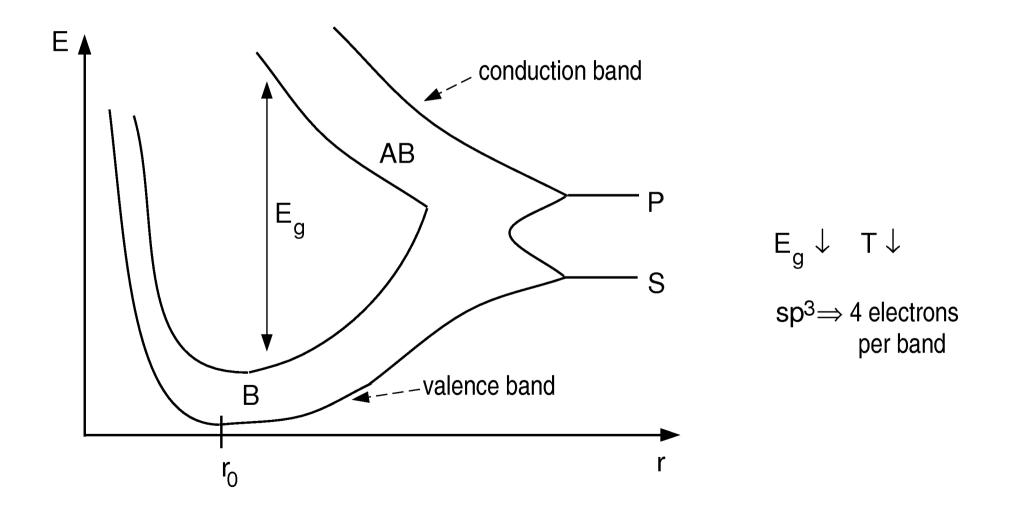
#### **Temperature Dependence of Band Gaps**

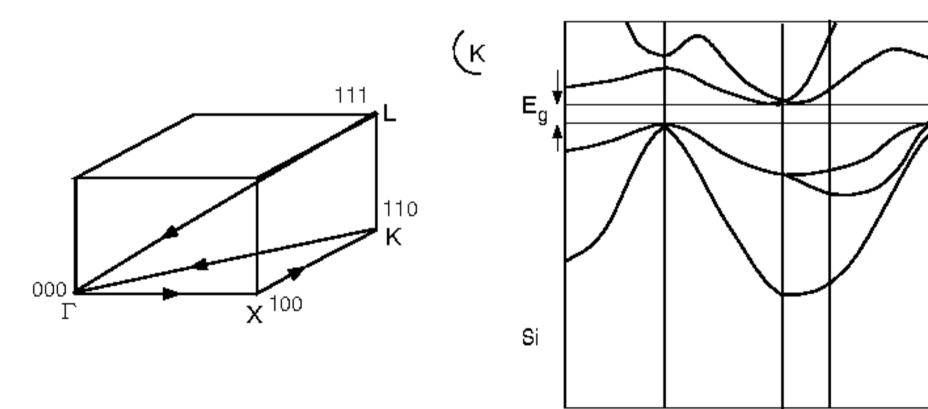
$$E_{g}(T) = E_{g}(0) - \frac{\alpha T^{2}}{T + \beta}$$



Increase of the inter-atomic spacing leads to the decrease of potential

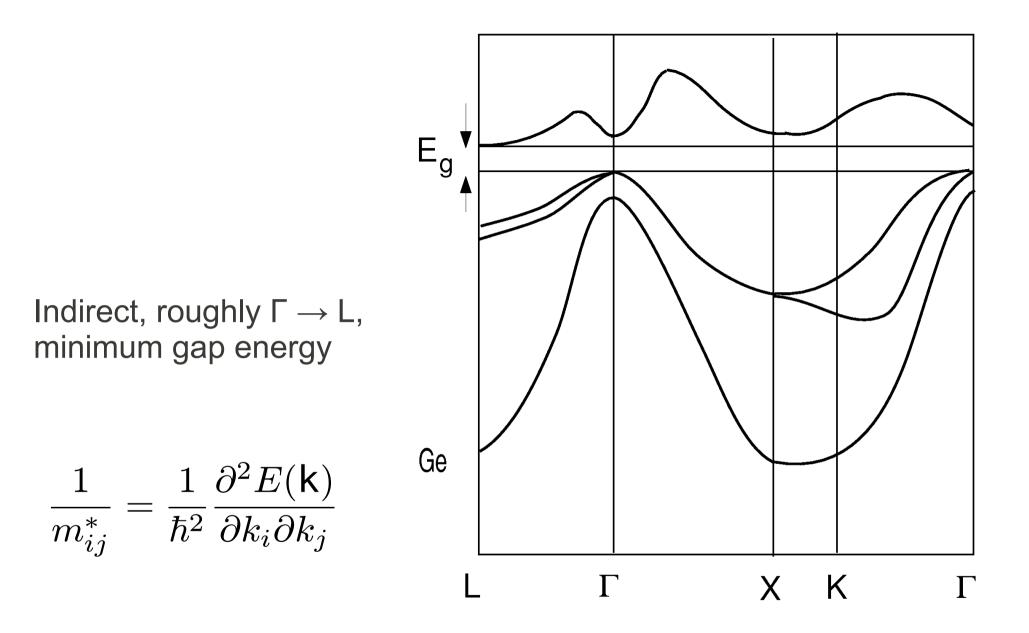
## Sketch of the sp<sup>3</sup> bands in Si vs. Si-Si separation





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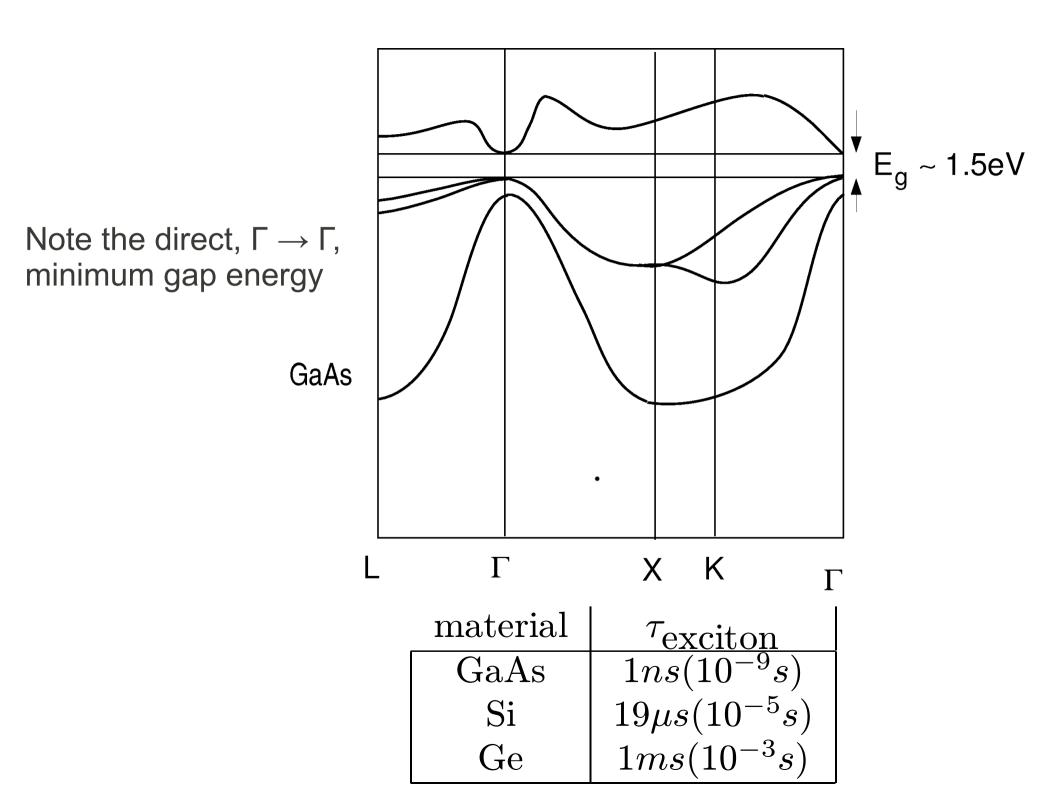
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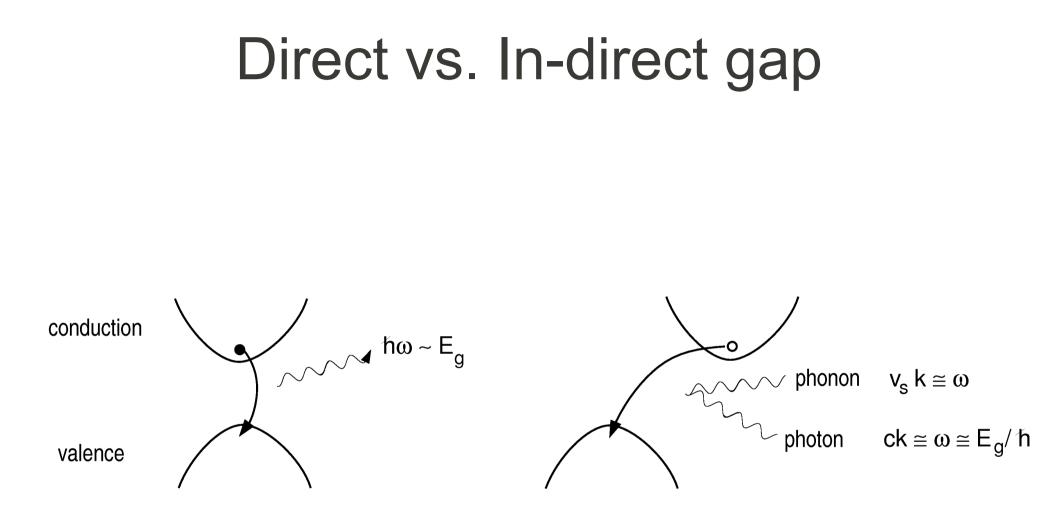


H <sup>1</sup>	<sup>1</sup> Periodic Table of the Elements © www.elementsdatabase.com											<sup>2</sup> He						
3 Li	4 ■ hydrogen Be = alkali metals ■ alkali earth metals						<ul> <li>poor metals</li> <li>nonmetals</li> <li>noble gases</li> </ul>						в <sup>5</sup>	C	N	0	۶ F	10 Ne
11 Na	12 Mg				netals			-	ases th me	tals			13 Al	14 <mark>Si</mark>	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 SC	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	; Zr	(	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 <mark>Sr</mark>	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	Cd		49 In	50 Sn	51 Sb	52 Te	53 	Xe Xe
Cs Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	Ηç		81 TI	82 Pb	8 Bi	84 P0	85 At	86 Rn
87 Fr	88 Ra	89 Ac				107 Uns	108 Uno											

Ce	Pr	60 Nd	61 Pm		64 Gd	<sup>65</sup> Tb			68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu			98 Cf	99 Es	100 Fm		102 No	

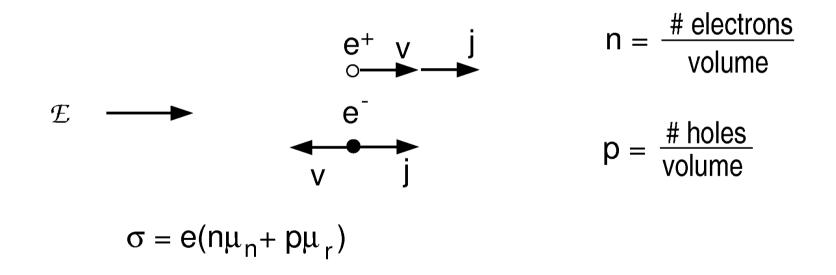
Most semiconductors have co-valiant bonding



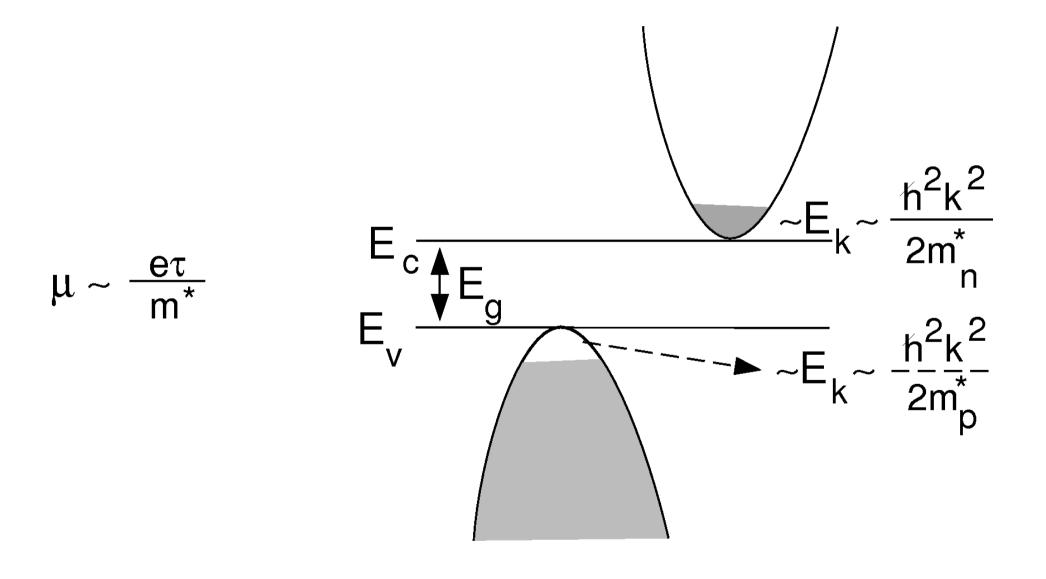


## Charge Carrier Density in Intrinsic Semiconductors

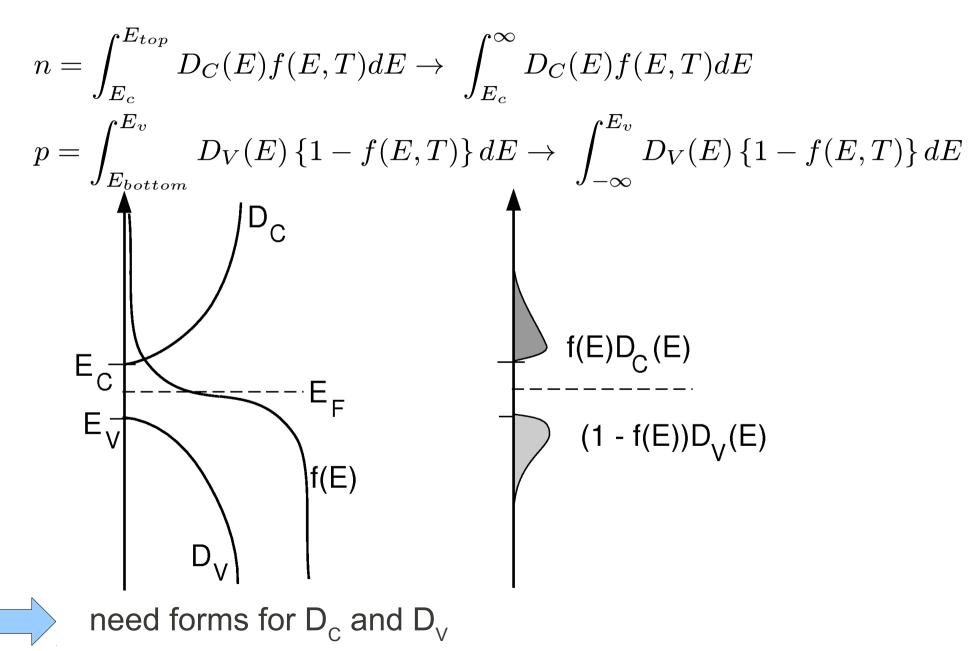
Both electrons and holes contribute to the conductivity



Mobilities are assumed to be constant: all of the conducting carriers are full near the top or bottom of bands, where  $E_k \sim \hbar^2 k^2/2m^*$  and the effective mass approximation is valid  $\mu \sim e\tau /m^*$ 



The carrier concentrations are highly T -dependent since all of the carriers in an intrinsic (un-doped) semiconductor are thermally induced (i.e. n = p = 0 at T = 0)



For the parabolic approximation  $E_k \simeq \frac{\hbar^2 \mathbf{k}^2}{2m^*}$  we have  $D(E) = \frac{(2m^*)^{\frac{3}{2}}}{2\pi^2 \hbar^3} \sqrt{E}$   $D_C(E) = \frac{(2m_n^*)^{\frac{3}{2}}}{2\pi^2 \hbar^3} \sqrt{E - E_C}$  $D_V(E) = \frac{(2m_p^*)^{\frac{3}{2}}}{2\pi^2 \hbar^3} \sqrt{E_V - E}$ 

Intrinsic (undoped) semiconductor n = p  $\longrightarrow$  EF must lie in the band gap.

However, if  $m_p^* \neq m_n^*$  (ie.  $D_c^* \neq D_v$ ) the chemical potential, EF, must be adjusted up or down from the center of the gap so that n = p.

Furthermore, the carriers which are induced across the gap are relatively high in energy, compared to  $k_B T$ , since typically  $E_g = E_c - E_v \gg k_B T$ .

	$E_g(eV)$	$n_i(cm^{-3})(300^{\circ}K)$
Ge	0.67	$2.4 imes 10^{13}$
Si	1.1	$1.5 imes\ 10^{10}$
GaAs	1.43	$5 imes 10^7$

 $\frac{1eV}{k_B} \sim 10000^{\circ}K \gg 300^{\circ}K \sim T$ Thus, assuming that  $E - E_F > \frac{E_g}{2} \gg k_BT$  $\frac{1}{e^{(E-E_F)/k_BT} + 1} \simeq \frac{1}{e^{(E-E_F)/k_BT}} = e^{-(E-E_F)/k_BT}$ 

**Boltzmann statistics** 

A similar relationship holds for holes where  $-(E - E_F) > \frac{E_g}{2} \gg k_B T$ 

$$1 - \frac{1}{e^{(E-E_F)/k_BT} + 1} \simeq 1 - \left\{1 - e^{-(E-E_F)/k_BT}\right\} = e^{-(E-E_F)/k_BT}$$

Since (1 - f(E)) = f(-E) and  $e^{(E - E_F)/k_B T}$  is small

$$\begin{array}{rcl} & & & & \\ & & & \\ & & & \\ & & & \\ & & = & \frac{\left(2m_n^*\right)^{\frac{3}{2}}}{2\pi^2\hbar^3}e^{E_F/k_BT}\int_{E_C}^{\infty}\sqrt{E-E_C}e^{-E/k_BT}dE \\ & & = & \frac{\left(2m_n^*\right)^{\frac{3}{2}}}{2\pi^2\hbar^3}\left(k_BT\right)^{\frac{3}{2}}e^{-\beta(E_C-E_F)}\int_0^{\infty}x^{\frac{1}{2}}e^{-x}dx \\ & & = & 2\left(\frac{2\pi m_n^*k_BT}{h^2}\right)^{\frac{3}{2}}e^{-\beta(E_C-E_F)} = N_{eff}^Ce^{-\beta(E_C-E_F)} \\ \\ & \text{Similarly } p = 2\left(\frac{2\pi m_p^*k_BT}{h^2}\right)^{\frac{3}{2}}e^{-\beta(E_V-E_F)} = N_{eff}^Ve^{-\beta(E_V-E_F)} \end{aligned}$$

In general, in the nondegenerate limit

$$np = 4\left(\frac{k_BT}{2\pi\hbar^2}\right)^3 \left(m_n^*m_p^*\right)^{\frac{3}{2}} e^{-\beta E_g}$$

Law of mass action - holds for both doped and intrinsic semiconductor so long as we remain in the nondegenerate limit.

For an intrinsic semiconductor, where n = p:

$$n_{i} = p_{i} = 2\left(\frac{k_{B}T}{2\pi\hbar^{2}}\right)^{\frac{3}{2}} \left(m_{n}^{*}m_{p}^{*}\right)^{\frac{3}{4}} e^{-\beta E_{g}/2}$$

However, we already have relationships for n and p involving  $E_c$  and  $E_v$  $n = p = N_{eff}^C e^{-\beta(E_C - E_F)} = N_{eff}^V e^{\beta(E_V - E_F)}$   $e^{2\beta E_F} = \frac{N_{eff}^V}{N_{eff}^C} e^{\beta(E_V + E_C)}$ 

$$E_{F} = \frac{1}{2}(E_{V} + E_{C}) + \frac{1}{2}k_{B}T\ln\left(\frac{N_{eff}^{V}}{N_{eff}^{C}}\right)$$
$$E_{F} = \frac{1}{2}(E_{V} + E_{C}) + \frac{3}{4}k_{B}T\ln\left(\frac{m_{p}^{*}}{m_{n}^{*}}\right)$$

Thus if  $m_p^* \neq m_n^*$ , the chemical potential EF in a semiconductor is temperature dependent.